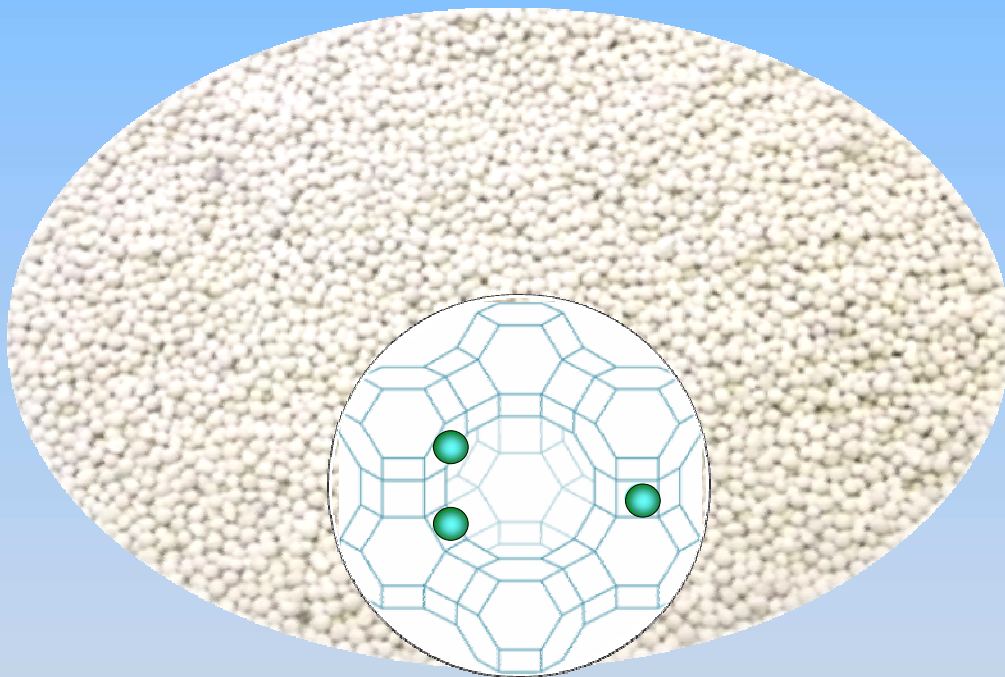


**Radioiodine Adsorbent  
for Nuclear Power Plant**

# **Silver Zeolite (AgX)**



**Rasa Industries, Ltd., Japan**

## **PREFACE**

Since the Fukushima-Daiichi nuclear accident, nuclear power plants in Japan start to improve safety countermeasures which are taken on the premise that an nuclear accident may happen. These countermeasures include installing and improving the filtered vents for preventing contamination of radioactive emission.

Rasa Industries, Ltd. has been providing the radioiodine absorbent since 1980, silver zeolite (AgX), which is highly efficient in removing radioactive iodine under the harsh conditions, especially organic iodine (CH<sub>3</sub>I). Currently, some Japanese nuclear power plants are utilizing or intending to utilize AgX filters as the countermeasure of radioactive iodine in their vent systems.

## **ISSUES OF FILTERED VENT**

1. Generation amount of organic iodine in radioactive emission was underestimated.
2. Hydrogen concentration in off-gas is high when vent starts to work.
3. Performances of adsorbent degrade due to the condensation of water vapor at the beginning of venting.
4. Traditional WET system cannot effectively remove radioactive organic iodine.
5. A high decontamination factor (DF) of 1000 should be achieved in eliminating organic iodine.

## **ISSUES OF SGTS AND ANNULUS**

The Fukushima nuclear accident revealed a number of issues in SGTS performances, such as hydrogen generation and explosion, temperature rise, and operation break-down in the case of power loss. It is absolutely necessary to strengthen the safety measures against organic iodine and hydrogen.

## **PRODUCT FEATURES, ADVANTAGES and BENEFITS**

1. AgX exhibits high adsorption efficiency of radioiodine, especially organic iodine even under high temperature, high humidity, and high pressure (please refer Tables 2). Furthermore, it also exhibits excellent adsorption characteristics in different gas compositions and temperatures (please refer Table 5). In the range of examined temperature, the higher the temperature, the higher the absorption efficiency.
2. Adsorption capacity for radioactive organic iodine is in the range of 85-200 mg/g. Using AgX filters can enhance the adsorption capability and reduce the waste.
3. High adsorption efficiencies can be kept under following conditions.
  - 1) within 0.06 sec. in residence time; 2) after heat treatment at 500 °C for over 10 days; 3) after high temperature steam exposure (over 100 °C) for over 10 days; 4) in gas mixtures containing hydrogen.
4. A compacted design of filter vent is possible. This is thought to be useful in anti-earthquake and preventing terrorism activities.
5. AgX has the effect of removing hydrogen (please refer to Table 6). This behavior can reduce the risk of hydrogen explosion.
6. Retention is over 99.9%. Adsorbed radioiodine can be retained almost completely.
7. Long product life cycle. There is hardly any aging deterioration under refrigeration or inert gas condition.
8. AgX is very stable in water. Even if it gets wet or is immersed in water, its adsorption performance can be recovered by drying it.
9. AgX filters can work without power supply.
10. Non-flammability. Secondary disasters such as fire will not occur.

## **PRODUCT APPLICATIONS**

1. As radioiodine collection filters used in WET and DRY systems.
2. As a substitute for activated carbon in SGTS of BWR-type reactor (please refer to the evaluation results in Tables 3 and 4).
3. As a substitute for activated carbon in annulus of PWR-type reactor.
4. As a radioiodine filter used in the central control room for reducing the risk of occupational exposures. Also it can be utilized together with HEPA and metal filters.
5. As radioiodine filters utilized in other nuclear facilities such as reprocessing plants, place of fuel handling and storage of nuclear fuel.

## PRODUCT EVALUATIONS

Table 1 Dependences of absorption efficiency of CH<sub>3</sub>I on the temperature of superheated steam. The evaluation was performed by TÜV SÜD.

| Bed depth (mm) | Residence time (sec.) | Absorption efficiency (%) |                     |                     |                      |                      |
|----------------|-----------------------|---------------------------|---------------------|---------------------|----------------------|----------------------|
|                |                       | 99 °C<br>(DPD* 0 K)       | 101 °C<br>(DPD 2 K) | 104 °C<br>(DPD 5 K) | 109 °C<br>(DPD 10 K) | 114 °C<br>(DPD 15 K) |
| 50.8           | 0.16                  | 99.860                    | 99.922              | 99.913              | 99.964               | 99.990               |
| 76.2           | 0.24                  | 99.988                    | 99.995              | 99.974              | 99.990               | 99.998               |
| 101.6          | 0.32                  | 99.997                    | 99.999              | 99.989              | 99.999               | 99.999               |

\*DPD: Dew Point Distance.

Other test conditions. Radioiodine: CH<sub>3</sub>I (I-131); gas comp.: steam/air=95/5 (super heated gas); P=0.98 bar.

Table 2 Relationships between absorption efficiency of CH<sub>3</sub>I and bed depth at the high temperature, relative humidity (RH) and pressure. The evaluation was performed by NUCON International Inc.

| Bed depth (mm)               | Residence time (sec.) | Absorption efficiency of CH <sub>3</sub> I (%) |
|------------------------------|-----------------------|--|
| T=130 °C; RH=95 %; P=399 kPa |                       |  |
| 50.8                         | 0.246                 | 99.967   |
| 76.2                         | 0.369                 | > 99.999                                       |
| 101.6                        | 0.492                 | > 99.999                                       |

Other test conditions. Linear velocity (LV)=20 cm/sec.; Radioiodine: CH<sub>3</sub>I (I-131).

Table 3 Absorption efficiencies of CH<sub>3</sub>I on AgX at different temperatures and relative humidity. The evaluation was performed by NUCON International Inc.

| Bed depth (mm) | Residence time (sec.) | Absorption efficiency of CH <sub>3</sub> I (%) |        |        |          |
|----------------|-----------------------|--|--------|--------|----------|
|                |                       | RH 95%   |        |        | RH 70%   |
|                |                       | 30 °C  | 60 °C  | 90 °C  | 66 °C    |
| 50.8           | 0.250                 | 98.738   | 99.685 | 99.970 | > 99.999 |
| 76.2           | 0.375                 | 99.850   | 99.950 | 99.983 | > 99.999 |
| 101.6          | 0.500                 | 99.962   | 99.987 | 99.995 | > 99.999 |

Other test conditions. P=103 kPa; LV=20.3 cm/sec.; Radioiodine: CH<sub>3</sub>I (I-131).

Table 4 Absorption efficiencies of elemental iodine (I<sub>2</sub>) on AgX at different bed depth/residence time. The evaluation was performed by NUCON International Inc.

| Bed depth (mm) | Residence time (sec.) | Absorption efficiency of I <sub>2</sub> (%) |
|----------------|-----------------------|---|
| 50.8           | 0.250                 | > 99.999                                    |
| 76.2           | 0.375                 | > 99.999                                    |
| 101.6          | 0.500                 | > 99.999                                    |

Other test conditions. T=66 °C; P=103 kPa; LV=20.3 cm/sec.; Radioiodine: I<sub>2</sub> (I-131).

Table 5 Adsorption efficiencies of CH<sub>3</sub>I on AgX in different periods and gas compositions. These conditions are corresponding to the beginning of a vent in BWR. The evaluation was performed by Rasa Industries, Ltd.

| Time (min.) | Gas composition (Vol. %) |                |       | AgX temp. (°C) | Adsorption efficiency (%)* |
|-------------|--------------------------|----------------|-------|----------------|----------------------------|
|             | H <sub>2</sub>           | N <sub>2</sub> | Steam |                |                            |
| 0 - 3       | 28                       | 60             | 12    | 22 - 70        | > 99.6                     |
| 3 - 6       | 28                       | 60             | 12    | 70 - 75        | > 99.6                     |
| 6 - 9       | 28                       | 60             | 12    | 75 - 75        | > 99.6                     |
| 15 - 18     | 23                       | 53             | 24    | 105 - 115      | > 99.6                     |
| 35 - 38     | 5                        | 12             | 83    | 140 - 145      | > 99.8                     |

\* Detection limit.

Other test conditions. Gas temp.=150 °C; residence time=0.18-0.21 sec.

Table 6 Variations of hydrogen content before and after the gas mixture (air + H<sub>2</sub>) flows through AgX. The evaluation was performed by Rasa Industries, Ltd.

| Air+H <sub>2</sub> conditions |                               |                       |                                       | AgX temp. (°C) | Results              |  |
|-------------------------------|-------------------------------|-----------------------|---------------------------------------|----------------|----------------------|--|
| Humid air* flow (ml/min.)     | H <sub>2</sub> flow (ml/min.) | Residence time (sec.) | Inlet H <sub>2</sub> content (vol. %) |                | Rising of temp. (°C) | Outlet H <sub>2</sub> content (vol. %) |
| 6600                          | 205                           | 0.87                  | 3.0                                   | 75             | 1                    | > 1.5                                  |
|                               |                               |                       |                                       | 120            | 15                   | < 0.5                                  |
|                               |                               |                       |                                       | 136            | 17                   | < 0.5                                  |

\*Humid air: air was bubbled in water at room temperature.

## PRODUCT SPECIFICATIONS

| No. | Items                    | Specifications      | Remarks                            |
|-----|--------------------------|---------------------|------------------------------------|
| 1   | Composition              | Synthesized zeolite |                                    |
| 2   | Metal cation of exchange | Silver (Ag)         |                                    |
| 3   | Exchange rate            | > 97 %              |                                    |
| 4   | Silver content           | > 36 %              | Dry base                           |
| 5   | Particle size            | 10 × 20 mesh        | JIS K 1474                         |
| 6   | Bulk density             | 1.2 g/ml (as is)    | JIS K 1474                         |
| 7   | Attrition                | < 3.0 %             | ASTM D-4058                        |
| 8   | Moisture content         | < wt 12 %           | 150 °C/3h dried                    |
| 9   | Pressure loss            | 0.15 kPa            | Thickness: 50 mm<br>LV: 20 cm/sec. |

## CONTACT INFORMATION

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**Rasa Island**

North Latitude: 24° 27' 57"  
East Longitude: 131° 11' 23"